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Re: Exhaust Gas Purifying Catalyst And Process For Purification Of Exhaust Gas
Applicant: Takeshi Matsumoto
Application No.: 10/535,331
Filing Date: 5/18/2005
Country: United States of America
Our Ref.: 66501-013US1

Number of pages incl. this cover sheet: 8

Attached please find the executed Declaration.

Please do not hesitate to contact me with any questions.

From:
Y. Rocky Tsao
Partner

TELEPHONE: (617) 500-2503
EMAIL: tsao@ortpatent.com

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In re Patent Application of
TAKESHI MATSUMOTO ET AL

5 Serial No. 10/535,331 Group Art Unit: 1755
Filed: May 18, 2005 Examiner: Wood, Elizabeth D
Title: EXHAUST GAS PURIFYING CATALYST AND PROCESS FOR
PURIFYING OF EXHAUST OF GAS

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**DECLARATION
UNDER 37 C.F.R. 1.132**

15 I, Takuji Nakane, a citizen of Japan, residing at
344-1-706, Waku, Aboshi-ku, Himeji-shi, Hyogo-ken, Japan,
declare:

I. I am one of co-inventors in the above referenced
application, and a chemist as well as a Researcher of AC
Research Laboratory of ICT CO., LTD., one of co-assignees
20 of this application, on the subject matters relating to this
application.

I graduated from Tokyo Institute of Technology, Faculty
of Science, Department of Chemistry in March 1997 and Graduate
School of Tokyo Institute of Technology, Department of
25 Chemistry in March 1999, and obtained a master degree majoring
surface science.

Since April, 1999, I have been the employee of ICT CO.,
LTD. at AC Research Laboratory and have been engaged in the
research work with respect to catalysts for purifying exhaust
30 gas of diesel engine.

I am well acquainted with all the other co-inventors
in this case, having worked with them on the development of

- 1 -

the invention described in this application.

II. In order to compare the effects of the catalysts of this application with those catalysts of Labarge et al I (US 6,489,259) and Labarge et al II (US 2002/0086793), I have conducted the following experiments:

Example 8

A catalyst was prepared by a similar method to Example 4 of this application. During the test for performance of purifying exhaust gas, light oil containing 0.03% by weight of sulfur was used as a fuel for an internal combustion engine. The results were summarized in Table 1A.

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TABLE 1A

Inlet temp. of catalyst layer	HC (ppm)	CO (ppm)	NOx (ppm)	O ₂ (vol.%)	CO ₂ (vol.%)	SO ₂ (ppm)	L.O. (mL/min)
500°C	30	90	380	6.5	10.2	12	8.5
450°C	45	100	360	8.1	9.1	-	7.0
400°C	50	110	320	9.6	8.0	-	6.5
350°C	60	120	240	11.0	7.0	-	5.5

L.O.: Light oil

Comparative Example 3

A similar method to Example 4 of this application was carried out except that ZSM-5 having 2-4 μm of particle diameter was used. Test results of Comparative Example 3 and Example 8 were summarized in Table 5.

TABLE 5 Initial NOx purification activity (%)

	500°C	450°C	400°C	350°C
Example 8	45	42	43	23
Comparative Example 3	35	35	37	22

Example 9

Catalyst was prepared by a similar method to Example 4 of this application except that 116g of ZSM-5, 91g of β -zeolite, 42g of Copper nitrate and 42g of silica sol were used. In this case, weight ratio of ZSM-5 and β -zeolite was 11:8. During the test for performance of purifying exhaust gas, light oil containing 0.003% by weight of sulfur as a fuel for an internal combustion engine. The results were summarized in Table 1B.

TABLE 1B

Inlet temp. of catalyst layer	HC (ppm)	CO (ppm)	NOx (ppm)	O ₂ (vol.%)	CO ₂ (vol.%)	SO ₂ (ppm)	L.O. (mL/min)
500°C	200	170	330	6.5	10.2	15	8.5
450°C	160	160	300	8.1	9.1	-	7.0
400°C	140	180	250	9.6	8.0	-	6.5
350°C	180	240	200	11.0	7.0	-	5.5

L.O.: Light oil

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Example 10

A similar method to Example 9 described above was carried out except that 160g of ZSM-5 and 43g of β -zeolite were used. In this case, weight ratio of ZSM-5 and β -zeolite was 5:1.

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Comparative Example 4

A catalyst was prepared by a similar method to Comparative Example 1, and a similar purification test was carried out.

The results of the test were shown in Table 2A including
5 those of Examples 9 and 10 and Comparative Example 4.

TABLE 2A

	CuO	ZSM5 (MFI zeolite)		BEA (β zeolite)	ZSM-5/ β -zeolite
		SiO ₂ /Si ₂ O ₃ =70	SiO ₂ /Si ₂ O ₃ =30		
		Average Crystal: less than 0.05 μ m	Average Crystal: 0.4 μ m		
Example 9	7	55		40	11:8
Example 10	7	76		19	5:1
Comparative Example 4	7	95			1:0

In Table 2: Unit: g/ litter catalyst

TABLE 6 Initial NOx purification activity (%)

	500°C	450°C	400°C	350°C
Example 9	44	45	44	33
Example 10	43	45	47	39
Comparative Example 4	42	46	47	28

TABLE 7 NOx purification activity (%) after durability test

	500°C	450°C	400°C	350°C
Example 9	42	43	41	20
Example 10	44	42	40	23
Comparative Example 4	43	39	40	15

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III. CONCLUSION

According to Table 5, a catalyst of this invention (i.e., the catalyst of Example 8), in which ZSM-5 had an average particle diameter not greater than 0.5 μm , was more active in reducing NOx in exhaust gas than a comparative catalyst (i.e., the catalyst of Comparative Example 3), in which ZSM-5 had an average particle diameter greater than 0.5 μm .

According to Tables 6 and 7, a catalyst of this invention (i.e., the catalyst of Example 9 or 10), in which the weight ratio of ZSM-5 to β zeolite fell within the range of 1:0.1 to 1:5 recited in claim 1, was more active in reducing NOx in exhaust gas at low inlet temperature, i.e., 350°C, than a comparative catalyst (i.e., the catalyst of Comparative Example 4), which, like LaBarge's catalyst, contained ZSM-5, but not β zeolite.

The undersigned Takuji Nakane declare that all the statements made herein are true; and further that these statements are made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Dated this *18th* day of February, 2008.

By Takuji Nakane
Takuji Nakane